# Radiation Safety Course (School of Science, the University of Tokyo)

# Biological Effects of Radiation to Human Body

Autumn-Winter 2021



#### 日本製鉄工場で社員2人被ばくか 年間限度 量の数十倍の可能性も

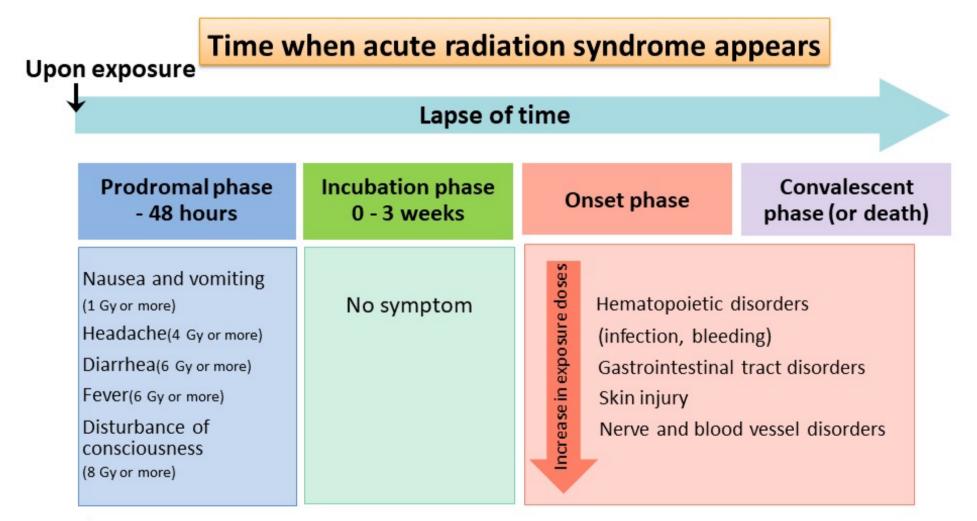
2021年6月11日 18時24分

兵庫県にある日本製鉄の工場で先月、エックス線を使う測定装置の点検中に事故が起き、 男性社員が年間の限度量の数倍から数十倍に及ぶ大量の被ばくをした可能性があることが 関係者への取材でわかりました。事故を重く見た厚生労働省は通知を出して同様の測定装 置を使っているほかの企業に被ばく防止の徹底を求めるとともに労働基準監督署などが事 故の状況を調べています。

During inspection, two employees at Nippon Steel Factory may have been exposed to radiation more than dozens of times the annual limit (50 mSv/y). The next day, they felt sick and went to see a doctor.

(Cited from NHK website)

#### **Acute Radiation Syndromes**

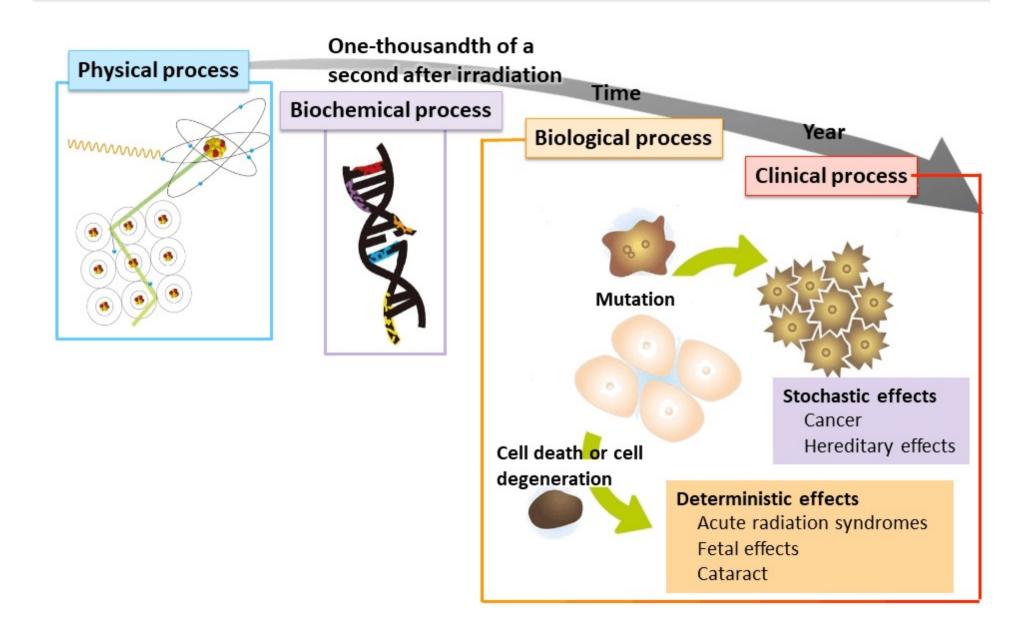


<sup>\*</sup> Acute radiation syndromes observed in the case of whole-body exposure to radiation exceeding 1 Gy (1,000 mGy) at one time

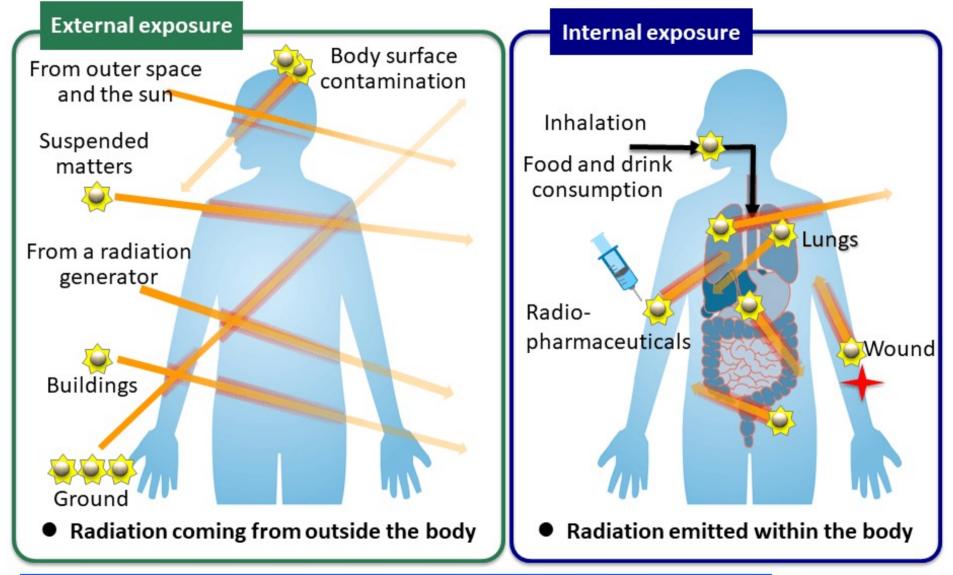
Gy: Grays

Source: "Basic Knowledge on Radiation" (a text for the Emergency Exposure Medical Treatment Training), Nuclear Safety Research Association

### Lapse of Time after Exposure and Effects G-4



#### **Internal and External Exposure**

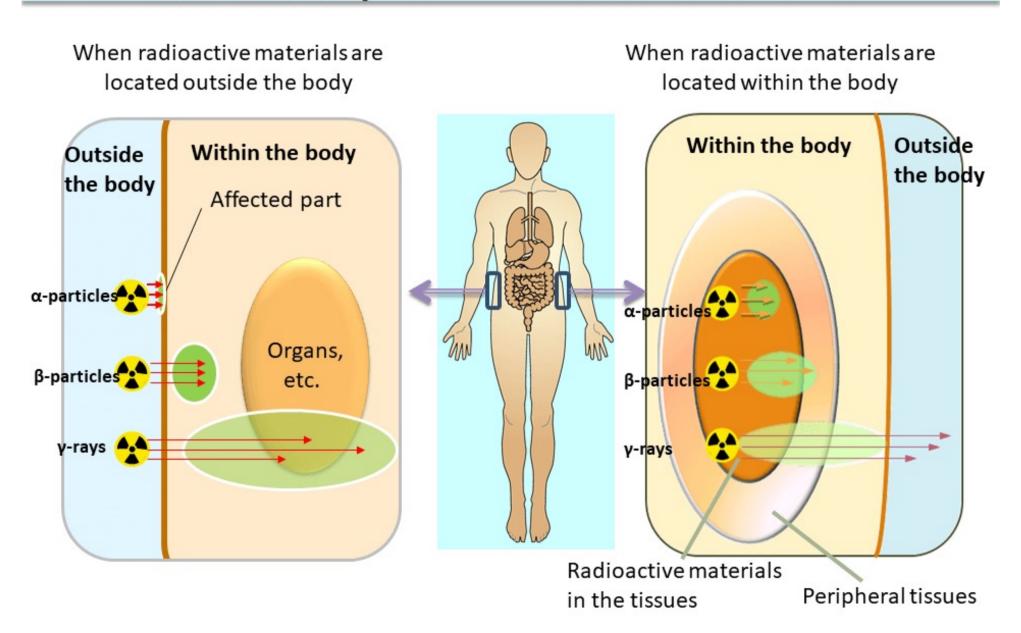


The body is equally exposed to radiation in both cases.



#### Penetrating Power and Range of Effects on the Human Body

G-6



#### **Internal Exposure**

#### (i) Ingestion

From the mouth (swallowing)
Absorption through the digestive tract

#### (ii) Inhalation

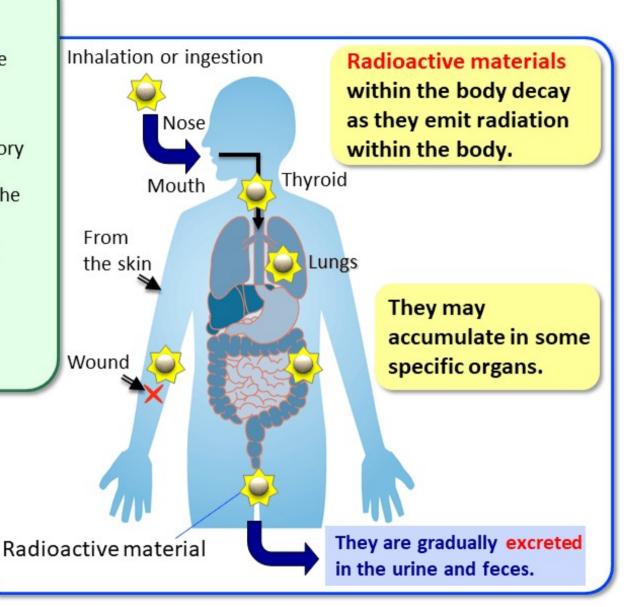
Incorporation from the respiratory airways
Absorption from the lungs and the surface of the airways

#### (iii) Percutaneous absorption

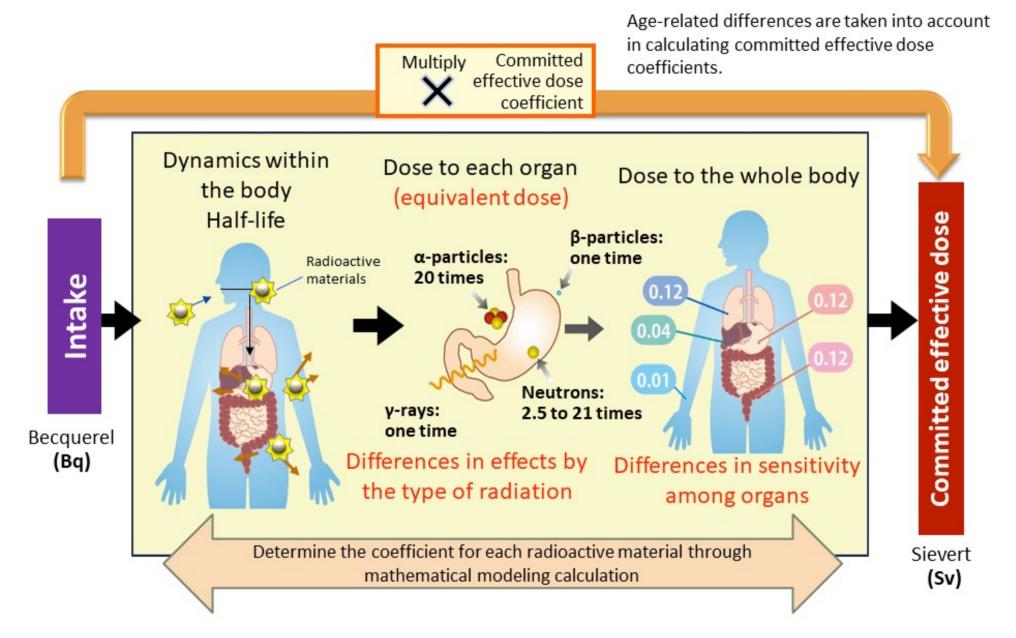
Absorption from the skin

#### (iv) Wound contamination

Contamination from a wound

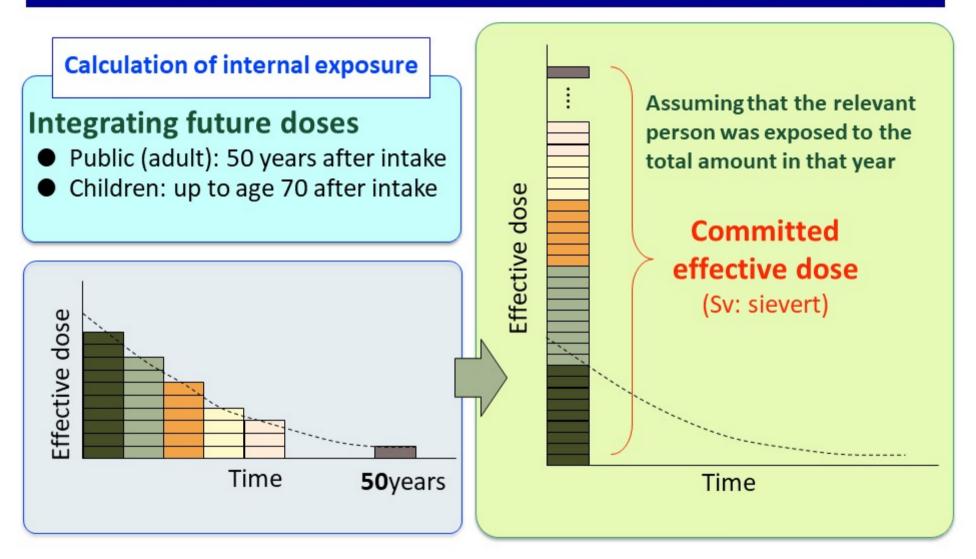


## Calculation of Internal Exposure Doses G-8



#### **Committed Effective Doses**

Exposure dose estimating how much radiation a person will be exposed to in lifetime from a single intake of radioactive materials



### **Conversion Factors to Effective Doses** G-10

#### Committed effective dose coefficients (µSv/Bq) (ingestion)

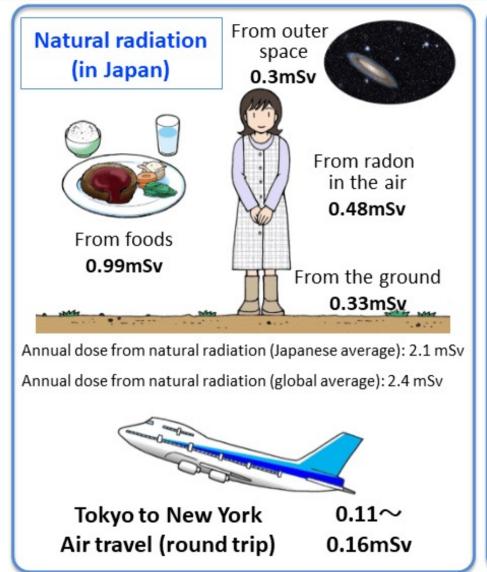
	Strontium-90	lodine-131	Cesium-134	Cesium-137	Plutonium-239	Tritium*
Three months old	0.23	0.18	0.026	0.021	4.2	0.000064
One year old	0.073	0.18	0.016	0.012	0.42	0.000048
Five years old	0.047	0.10	0.013	0.0096	0.33	0.000031
Ten years old	0.06	0.052	0.014	0.01	0.27	0.000023
Fifteen years old	0.08	0.034	0.019	0.013	0.24	0.000018
Adult	0.028	0.022	0.019	0.013	0.25	0.000018

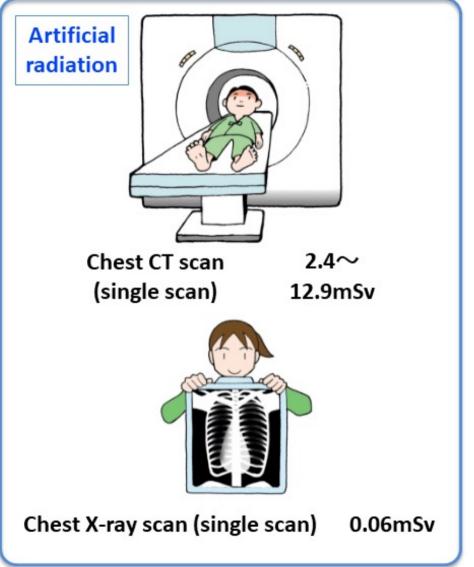
μSv/Bq: microsieverts/becquerel

\*Tissue free water tritium

Source: ICRP Publication 119, Compendium of Dose Coefficients based on ICRP Publication 60, 2012, International Commission on Radiological Protection (ICRP)

# Exposure Dose from Natural and Artificial Radiation G-11





mSv: millisieverts

Sources: Prepared based on the 2008 UNSCEAR (United Nations Scientific Committee on the Effects of Atomic Radiation) Report; and "Environmental Radiation in Daily Life (2011)," new edition, Nuclear Safety Research Association; ICRP (International Commission on Radiological Protection) 103, etc.

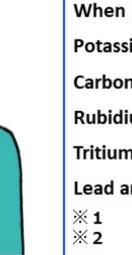
## **Breakdown of Natural Exposure Doses** G-12 (Japanese)

Type of exposure	Breakdown of radiation sources	Effective dose (mSv/year)
External	Cosmic rays	0.3
exposure	Ground radiation	0.33
	Radon-222 (indoors and outdoors)	0.37
Internal exposure	Radon-220 (thoron) (indoors and outdoors)	0.09
(inhalation)	Smoking (Lead-210, Polonium-210, etc.)	0.01
	Others (uranium, etc.)	0.006
	Mainly Lead-210 and Polonium-210	0.80
Internal exposure	Tritium	0.0000082
(ingestion)	Carbon-14	0.01
	Potassium-40	0.18
	2.1	

Source: "Environmental Radiation in Daily Life (2011)," Nuclear Safety Research Association Source: BOOKLET to Provide Basic Information Regarding Health Effects of Radiation

### Natural Radioactive Materials in the Body and Foods G-13

Radioactive materials in the body



Radioactivity concentrations (Potassium-40) in foods

When body weight is	60kg			
Potassium-40	<b>※ 1</b>	4,000Bq		
Carbon-14	<b>※ 2</b>	2,500Bq		
Rubidium-87	<b>※ 1</b>	500Bq		
Tritium	<b>※ 2</b>	100Bq		
Lead and polonium	Ж3	20Bq		
× 1 Nuclides originating from the Earth				

Nuclides originating from the Earth
 2 Nuclides derived from N-14 originating from

cosmic rays

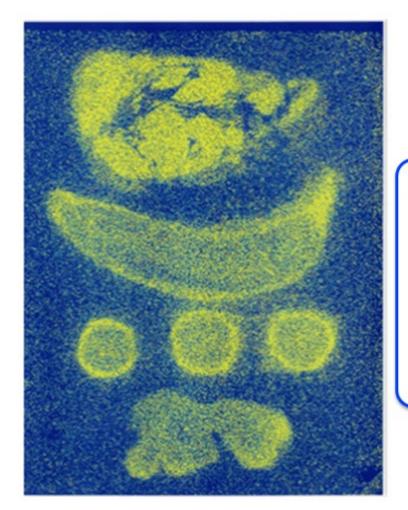
X 3 Nuclides of the uranium series originating from the Earth

Rice: 30; Milk: 50; Beef: 100; Fish: 100; Dry milk: 200; Spinach: 200;

Potato chips: 400; Green tea: 600; Dried shiitake: 700; Dried kelp: 2,000 (Bq/kg)

Bq: becquerels Bq/kg: becquerels/kilogram

#### **Visualized Radiation**



#### Radiation from foods

- · Mostly β-particles from Potassium-40
- The natural abundance ratio of Potassium-40\* is 0.012%.
- · Potassium-40 has a half-life of **1.26** × **10**<sup>9</sup> years.

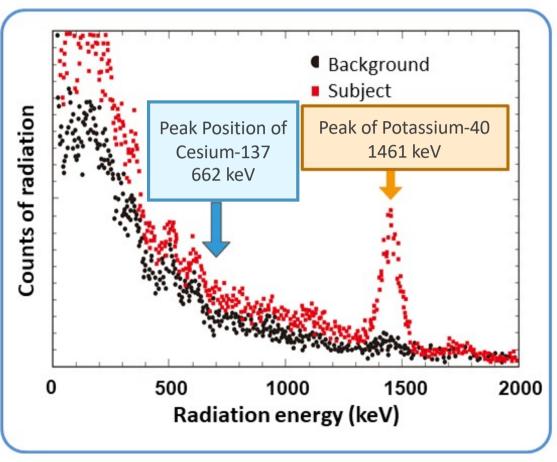
\*Percentage of Potassium-40 relative to the total amount of potassium found in nature

Radiographs of pork meat, banana (cut vertically and horizontally), and ginger

### Data on Internal Exposure Measured by G-15 Direct Counting



Whole-body counter



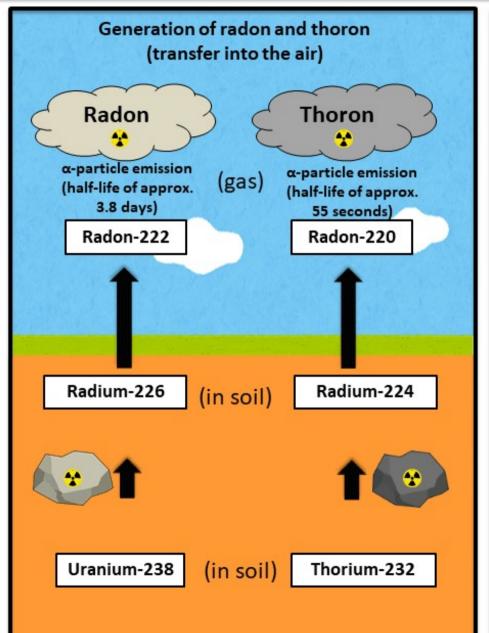
Measure radiation emitted from within the body ⇒ Measure internal radioactivity for each radioactive material

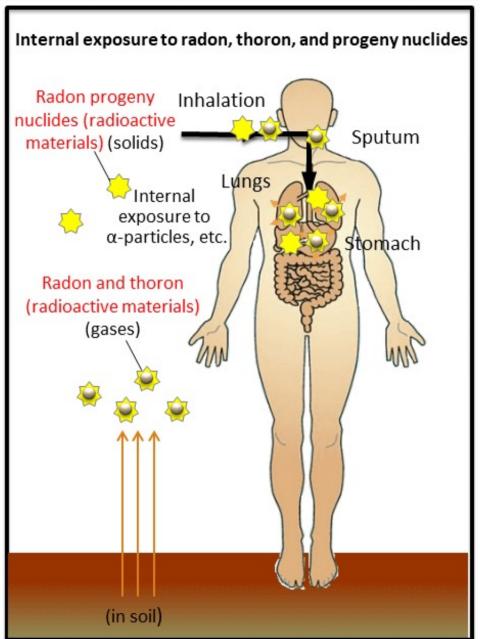
The amount of potassium in the body is around 2 g per 1 kg of body weight, and approx. 0.01% of that amount is radioactive potassium (K-40)

keV: kiloelectronvolts

Source: BOOKLET to Provide Basic Information Regarding Health Effects of Radiation (partially modified)

## Internal Exposure to Radon and Thoron through Inhalation G-16





### Question 2

An adult worker accidentally ingested 1 MBq of Strontium-90.

In this case, the committed effective dose is calculated to be...

OO mSv