

Radiation Safety Course

(School of Science, the University of Tokyo)

Biological Effects of Radiation to Human Body

FY2023

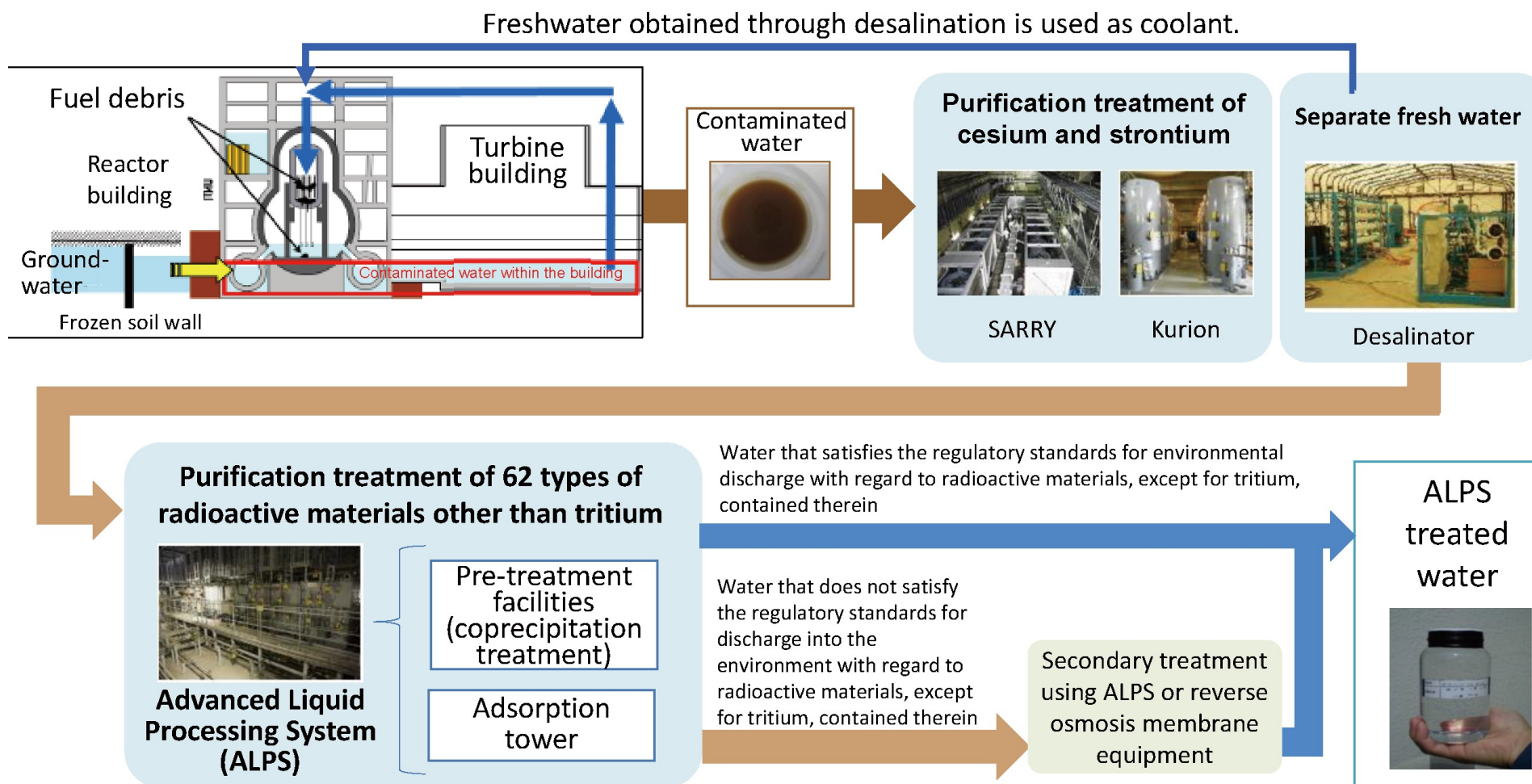


福島第一原発 処理水放出の開始時期 春から夏ごろ見込む 政府

2023年1月13日 18時10分

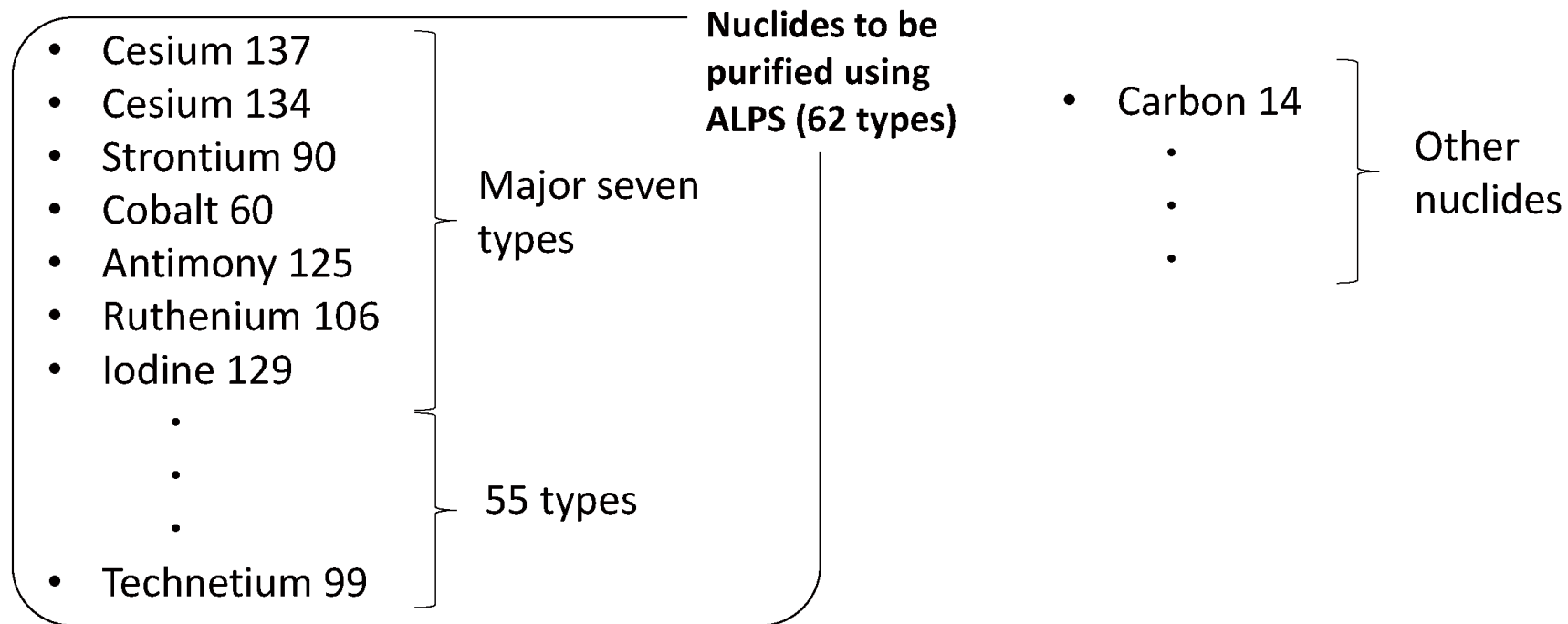
The Japanese government gave an estimation that the release of treated water from destroyed Fukushima nuclear power plant into the sea would be around this spring or summer.

- Contaminated water with radioactive materials is being generated after the accident at TEPCO's Fukushima Daiichi NPS. "ALPS treated water" refers to the water that has been treated by the Advanced Liquid Processing System (ALPS) and other equipment and has been purified to a level where contained radioactive materials, except for tritium, satisfy the regulatory standards for discharge into the environment.



(Source) Prepared based on "Fukushima Daiichi Nuclear Power Station: Contaminated water management: What is 'slurry'? Why is it generated? How is it stored?" by the Agency for Natural Resources and Energy (https://www.enecho.meti.go.jp/en/category/special/article/detail_157.html)

- Contaminated water generated at TEPCO's Fukushima Daiichi NPS not only contains tritium but also contains Cesium 137, Strontium 90 and other radioactive materials which are seldom detected in water discharged from ordinary nuclear power stations.
- Out of those radioactive materials, 62 types of nuclides that are likely to be contained in the contaminated water at certain levels in consideration of regulatory standards respectively set for those types of nuclides are purified by the use of the Advanced Liquid Processing System (ALPS) and other equipment to the extent that their concentrations become below those regulatory standards.



(Source) Prepared based on "Advanced Liquid Processing System (ALPS)" (https://www.tepco.co.jp/nu/fukushima-np/f1/genkyo/fp_cc/fp_alps/) (in Japanese) and "Performance test regarding secondary treatment of ALPS treated water," etc. by Tokyo Electric Power Company Holdings

トリチウム濃度の比較

60,000

ベクレル/L

Comparison of Tritium Concentrations

Bq/L

$\frac{1}{40}$

10,000

ベクレル/L

Bq/L

$\frac{1}{7}$

Less than
1500 Bq/L

1,500

ベクレル/L未満

国の
安全基準

Japan's safety
standards

WHO
飲料水基準

WHO's drinking
water standard

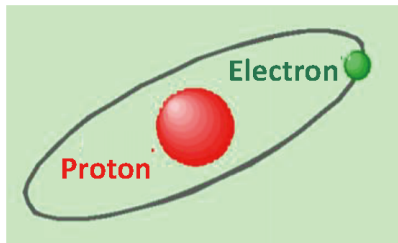
ALPS処理水を
海水放出する際の濃度

The concentration of ALPS-treated
water when it is released into the sea

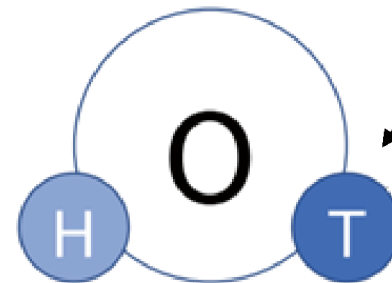
Tritium is a radioisotope of hydrogen, called "hydrogen-3."

【Structure of water molecules】

Ordinary hydrogen
(light hydrogen)



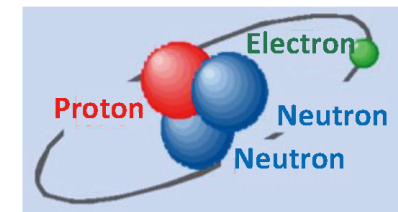
Water molecule solely
consisting of ordinary hydrogen



Water molecule consisting of
ordinary hydrogen and tritium



Tritium (hydrogen-3)

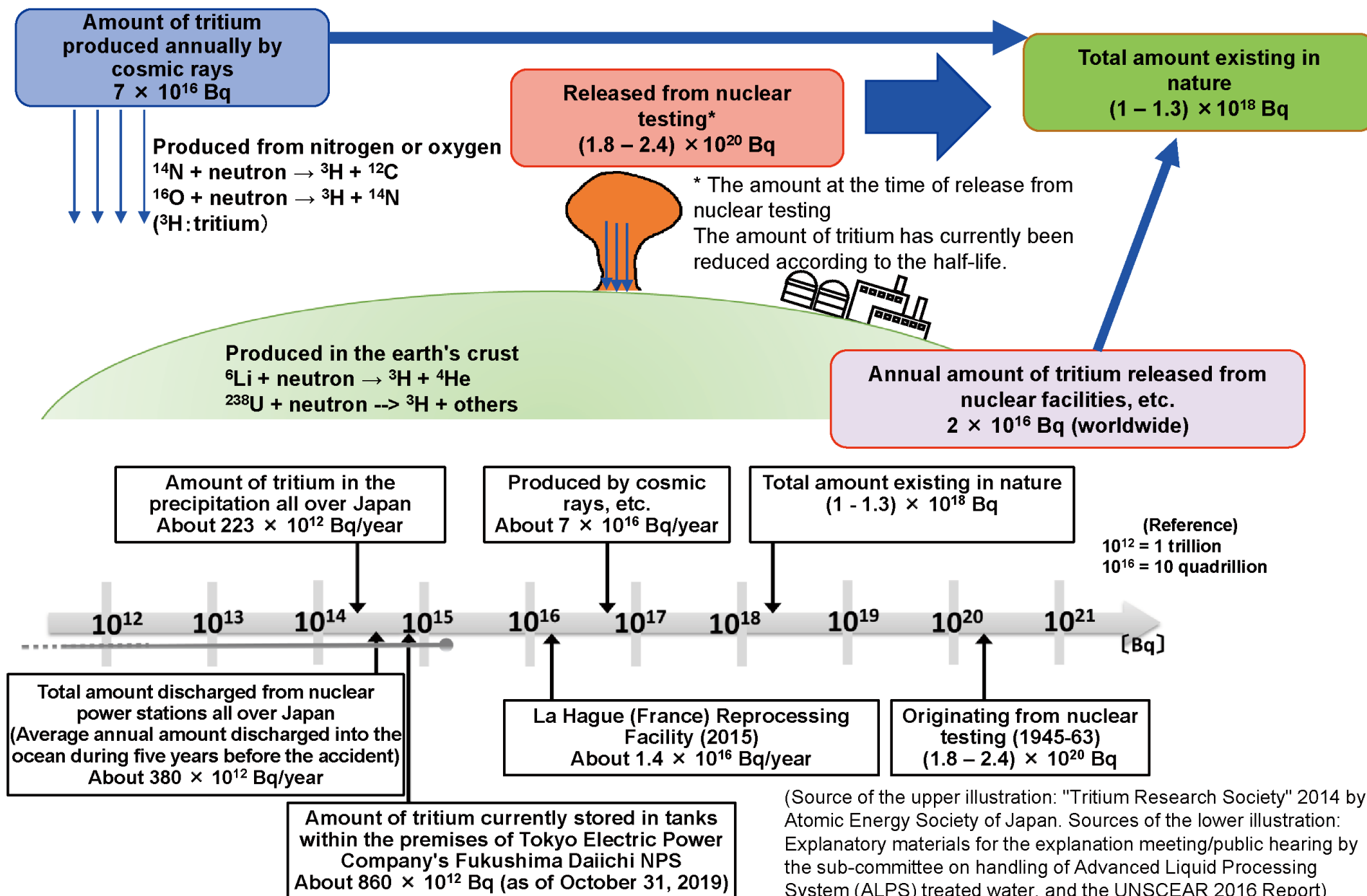


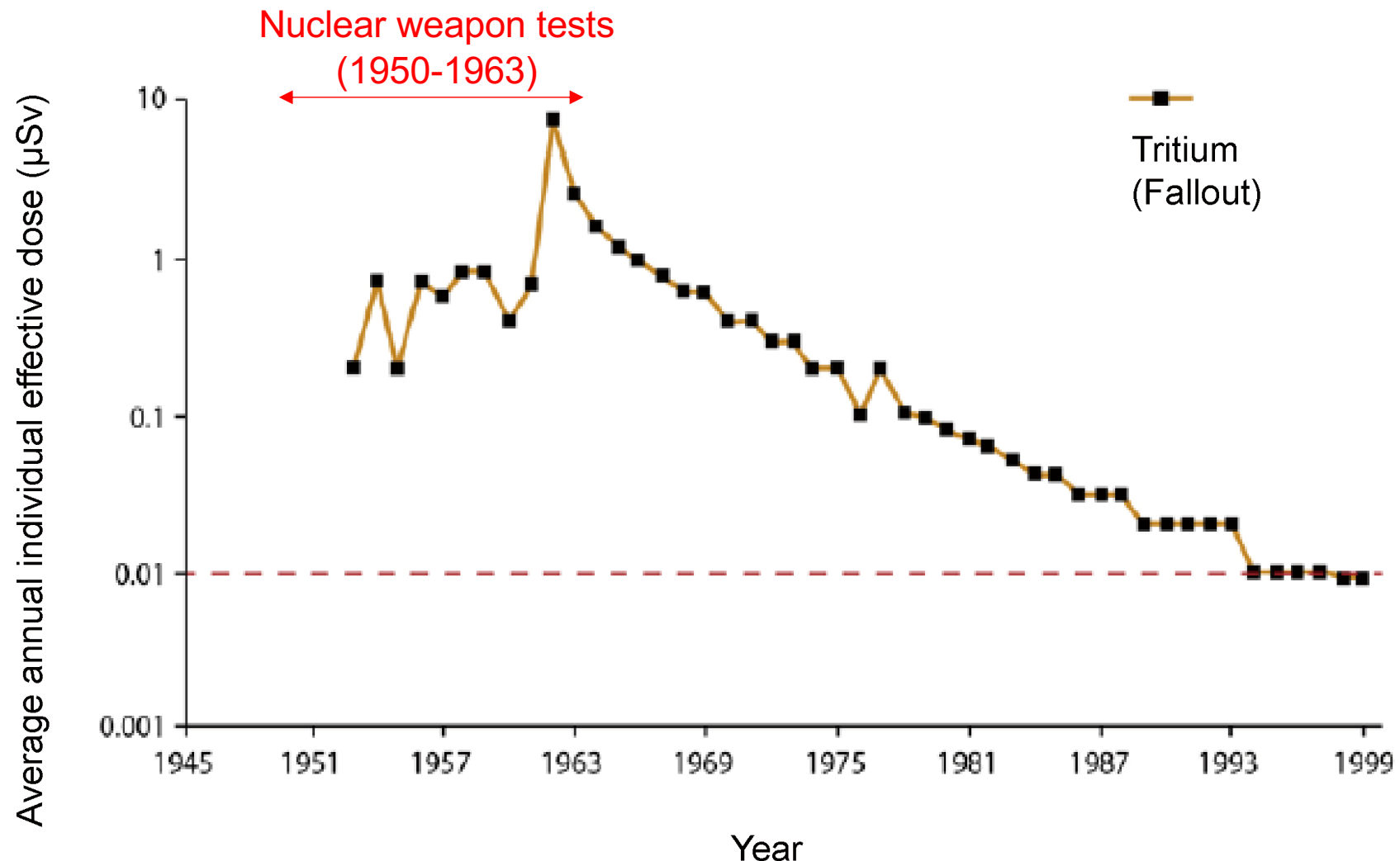
Physical half-life: 12.3 year

E_{mean} : 5.7 keV

E_{max} : 18.6 keV

Source: Prepared based on the "Important Stories on Decommissioning 2018" by the Agency for Natural Resources and Energy, METI, the "Tritiated Water Task Force Report" by the Tritiated Water Task Force (2016), and the "Scientific Characteristics of Tritium (draft)" by the Subcommittee on Handling of the ALPS Treated Water



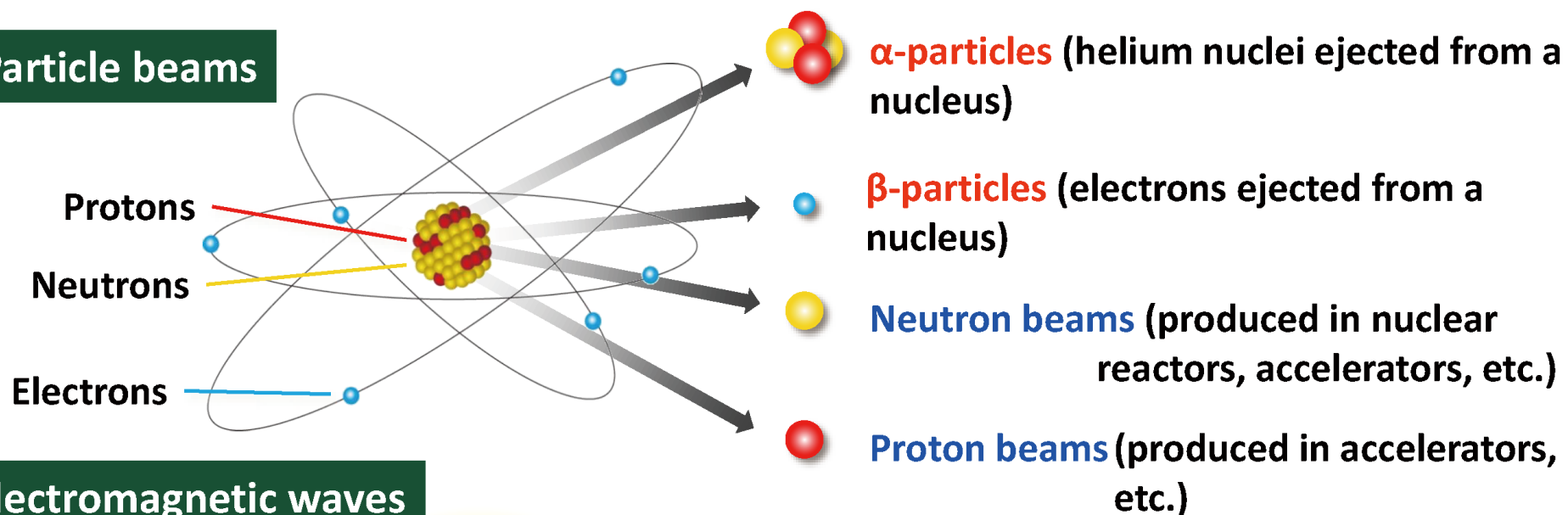
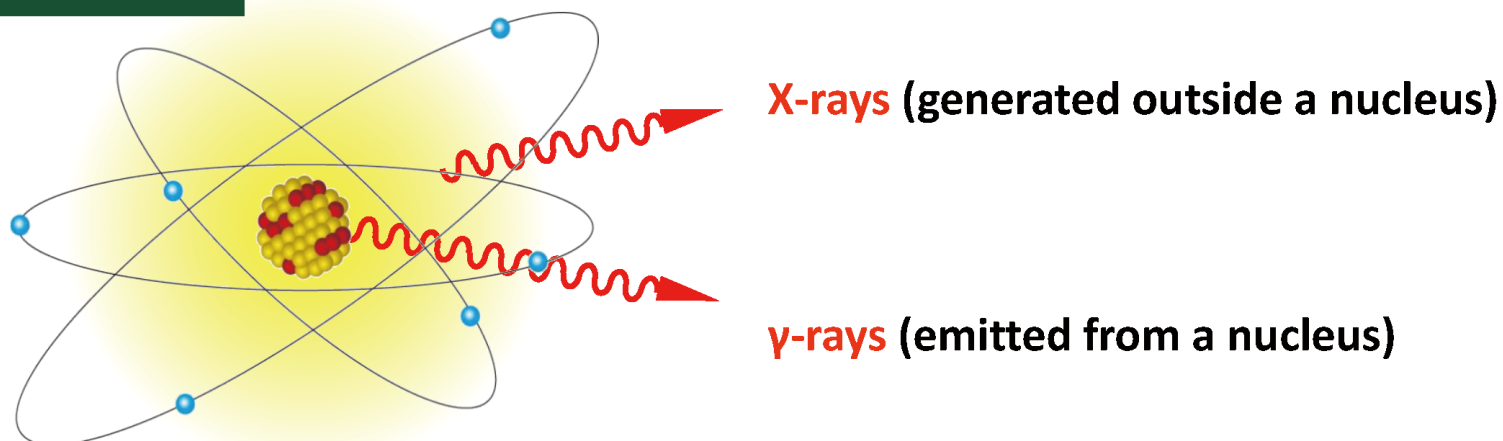


Source: UNSCEAR 2016 Report, Annex C-Biological effects of selected internal emitters-Tritium

Source: BOOKLET to Provide Basic Information Regarding Health Effects of Radiation

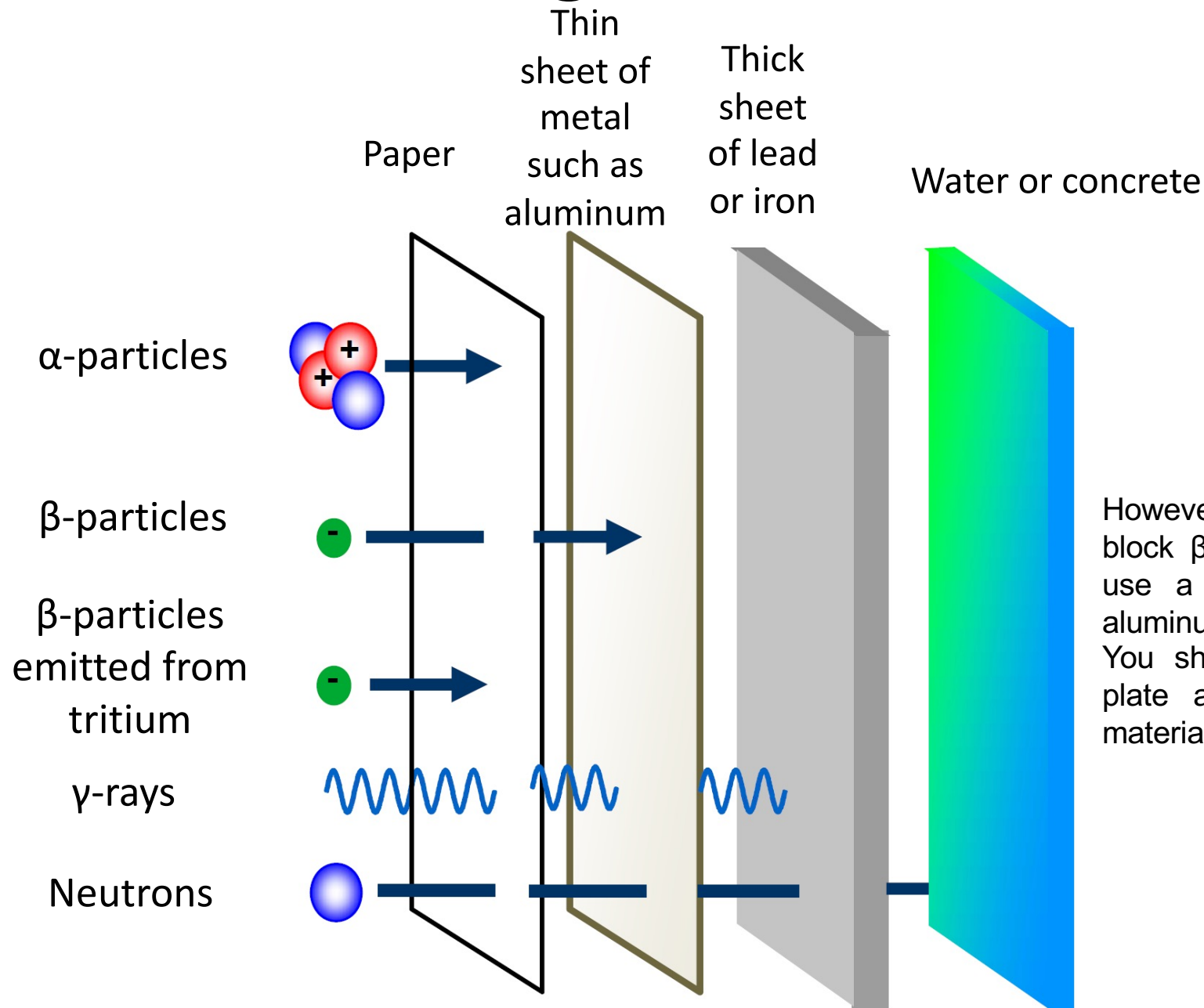
Ionizing radiation

Radiation that causes ionization

Particle beams**Electromagnetic waves**

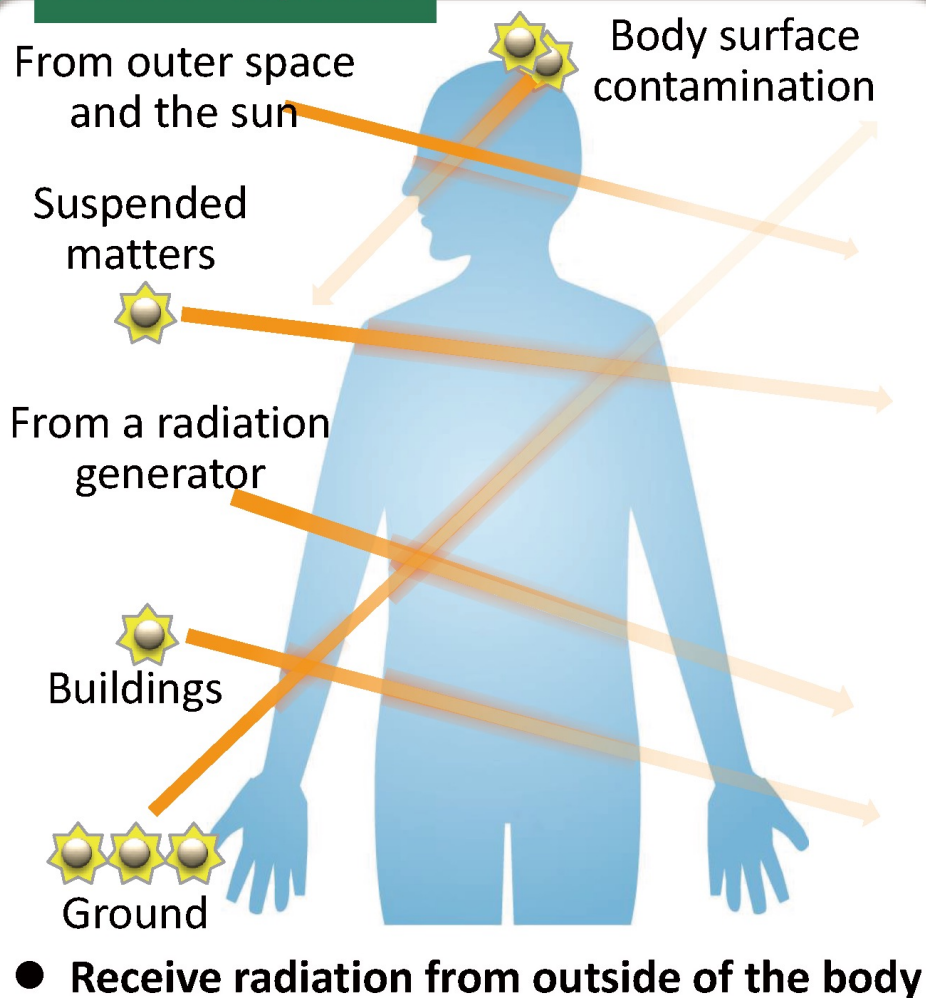
Penetrating Power of Radiation

G-10

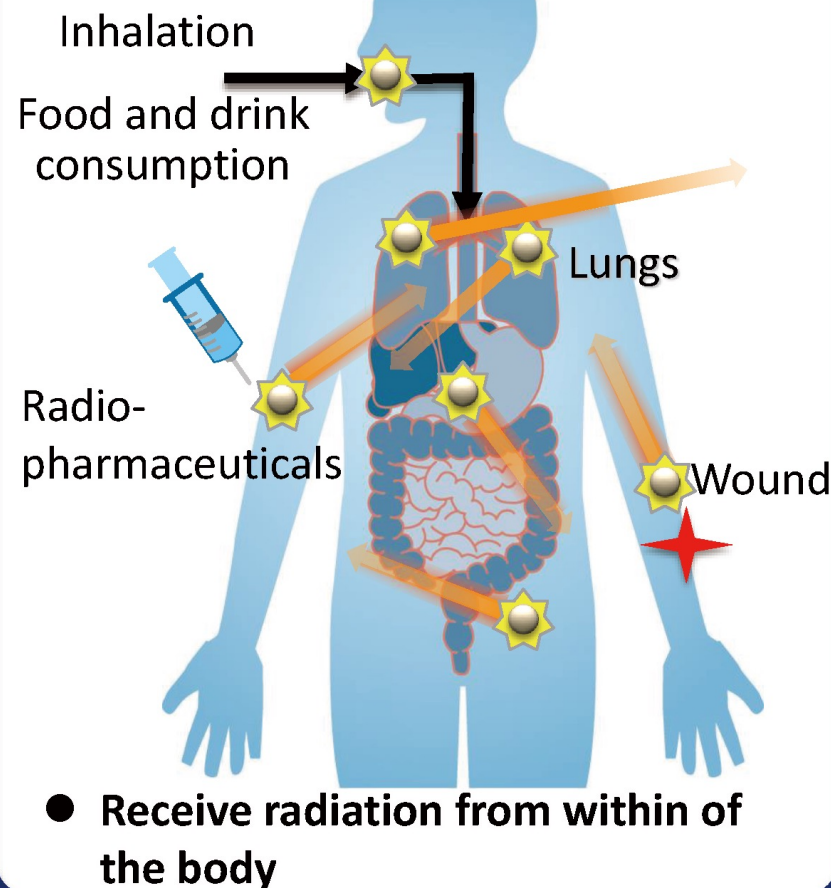


However, when you want to block β -particles, you should use a plastic sheet or an aluminum plate. You should not use a lead plate as the first shielding material against β -particles.

External exposure



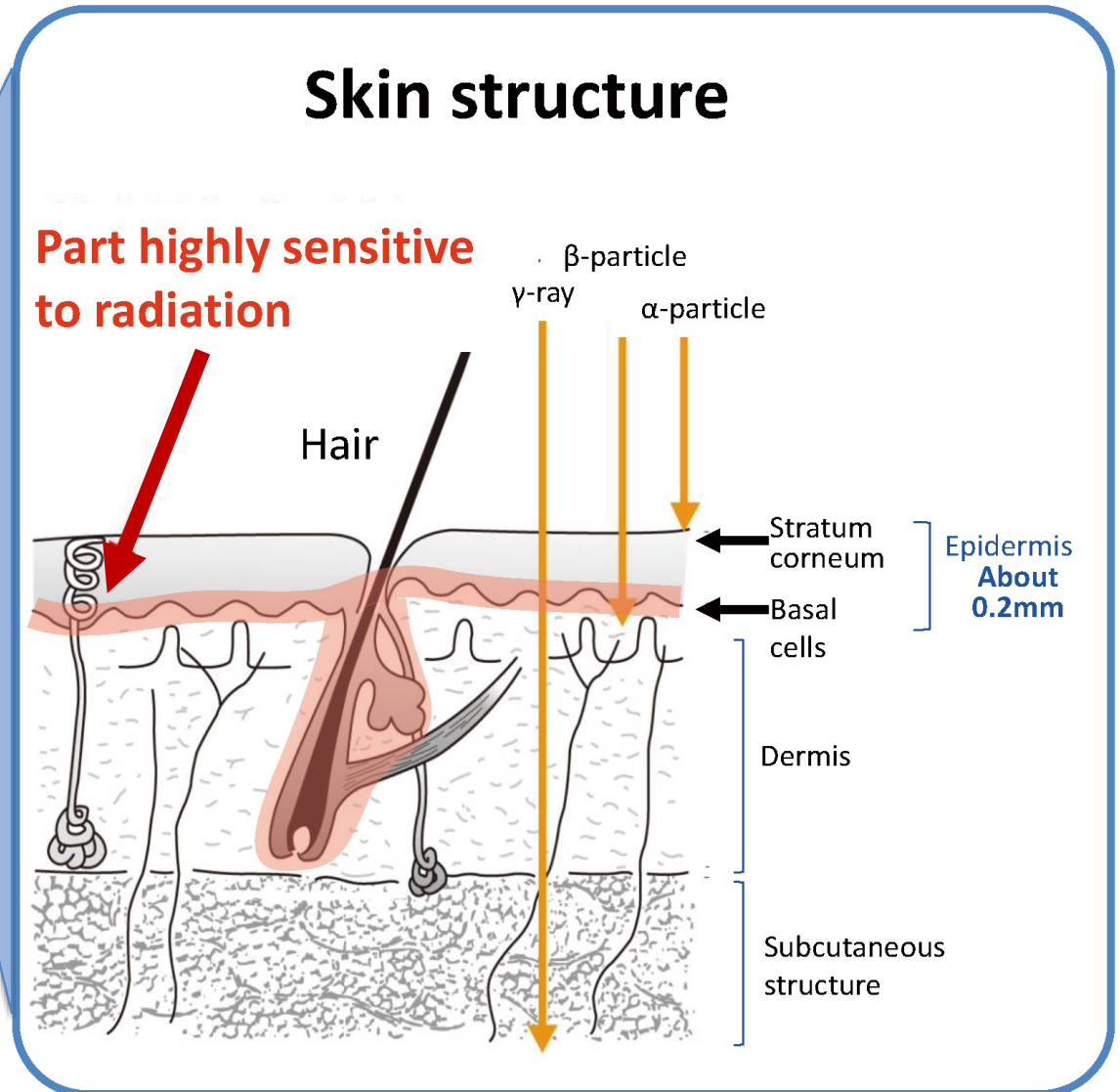
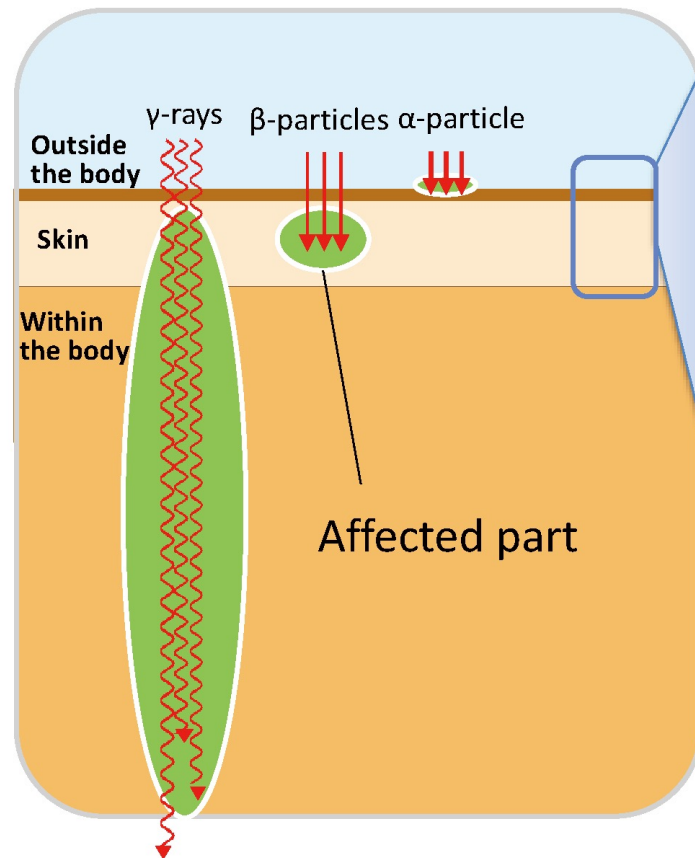
Internal exposure



The body is equally exposed to radiation in both cases.



Radioactive materials



(i) Ingestion

From the mouth (swallowing)
Absorption through the digestive tract

(ii) Inhalation

Incorporation from the respiratory airways
Absorption from the lungs and the surface of the airways

(iii) Percutaneous absorption

Absorption from the skin

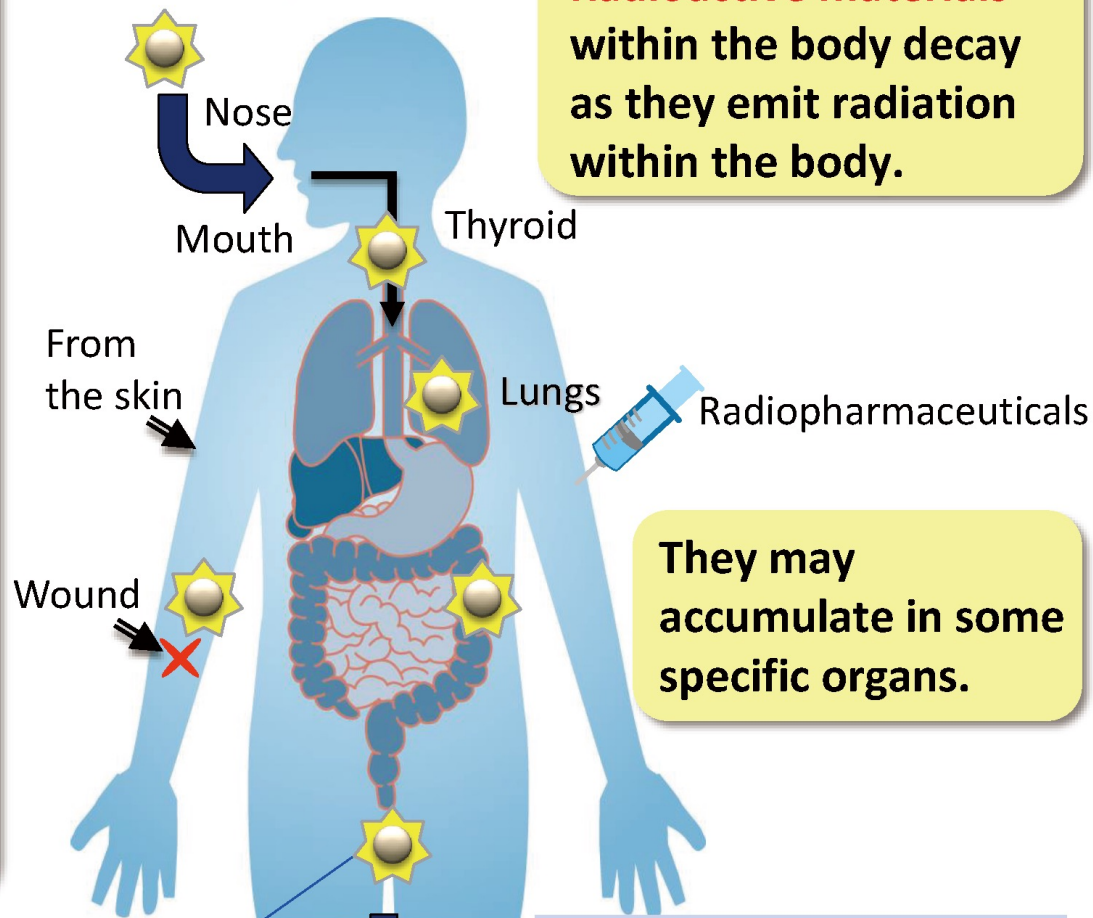
(iv) Wound contamination

Contamination from a wound

(v) Intake of radiopharmaceuticals

Injection, oral administration (→ (i))
Inhalation of gas (→ (ii))

Inhalation or ingestion



Radioactive materials within the body decay as they emit radiation within the body.

They may accumulate in some specific organs.

Radioactive material

They are gradually excreted in the urine and feces.

The characteristics of radioactive materials that especially cause problems in internal exposure

- (i) α -emitters $>$ β -emitters or γ -emitters
- (ii) Materials that enter easily but are difficult to excrete
- (iii) Materials that are likely to accumulate in specific organs

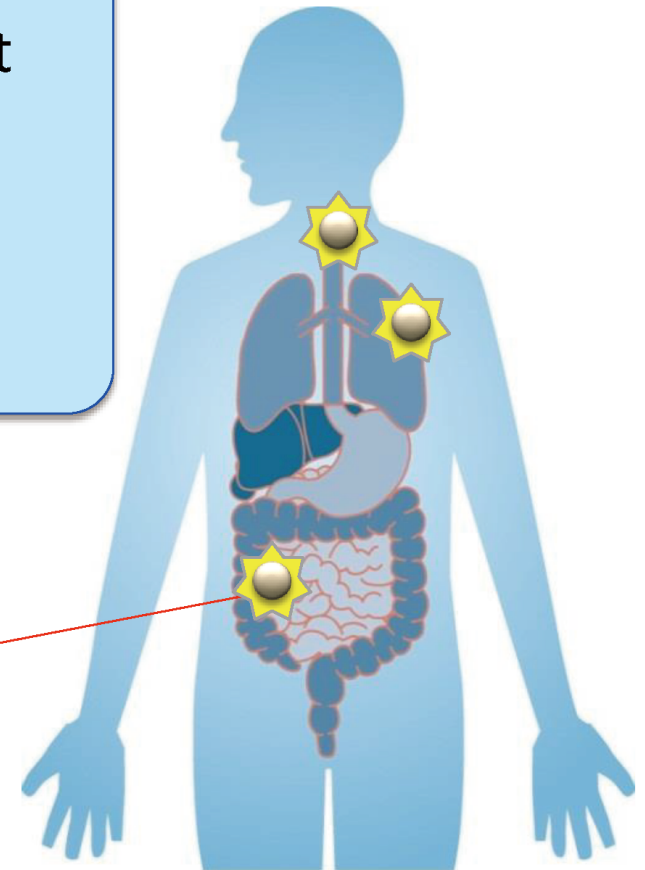
$$\frac{1}{T_e} = \frac{1}{T_p} + \frac{1}{T_b}$$

T_e : Effective half-life

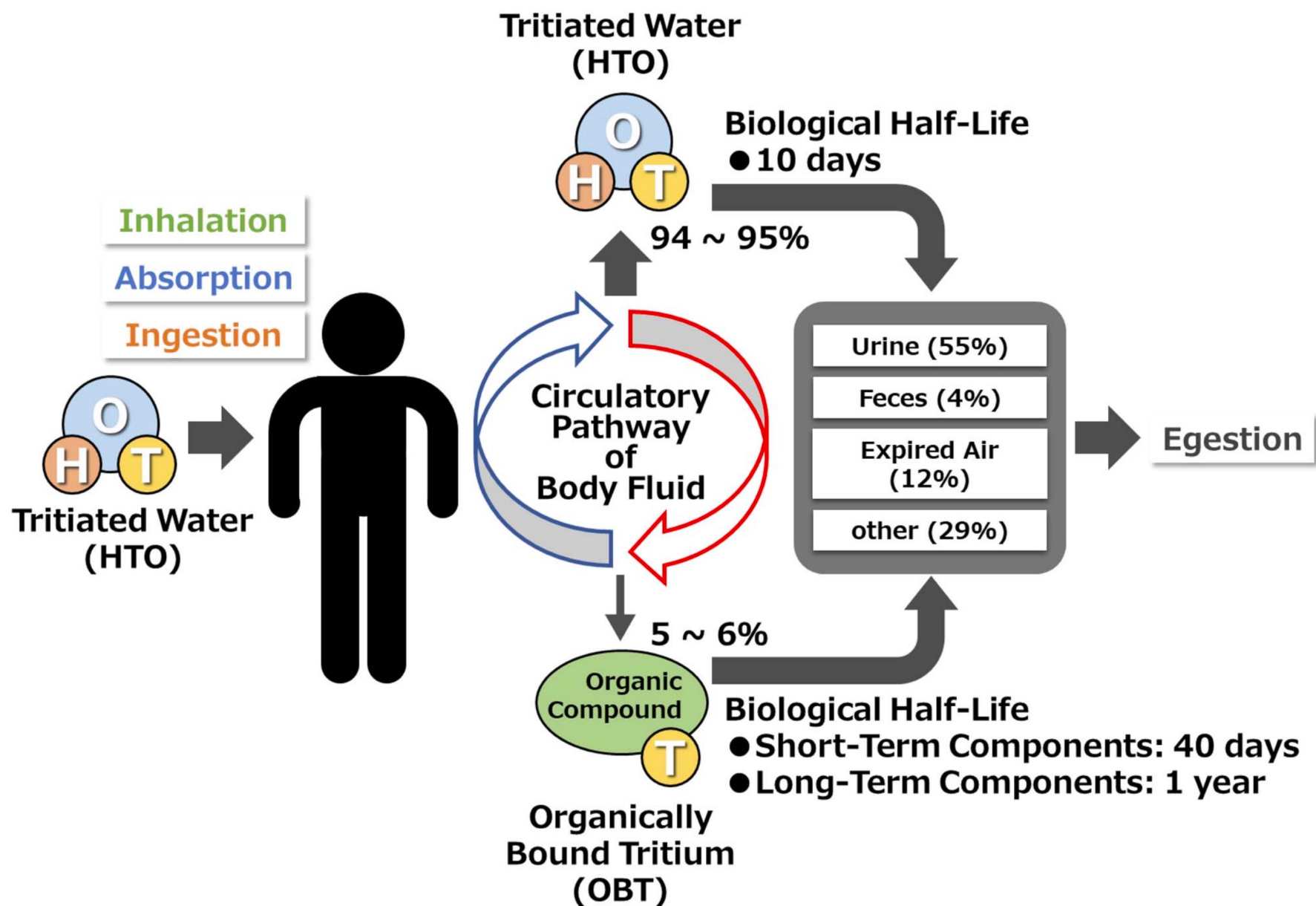
T_p : Physical half-life

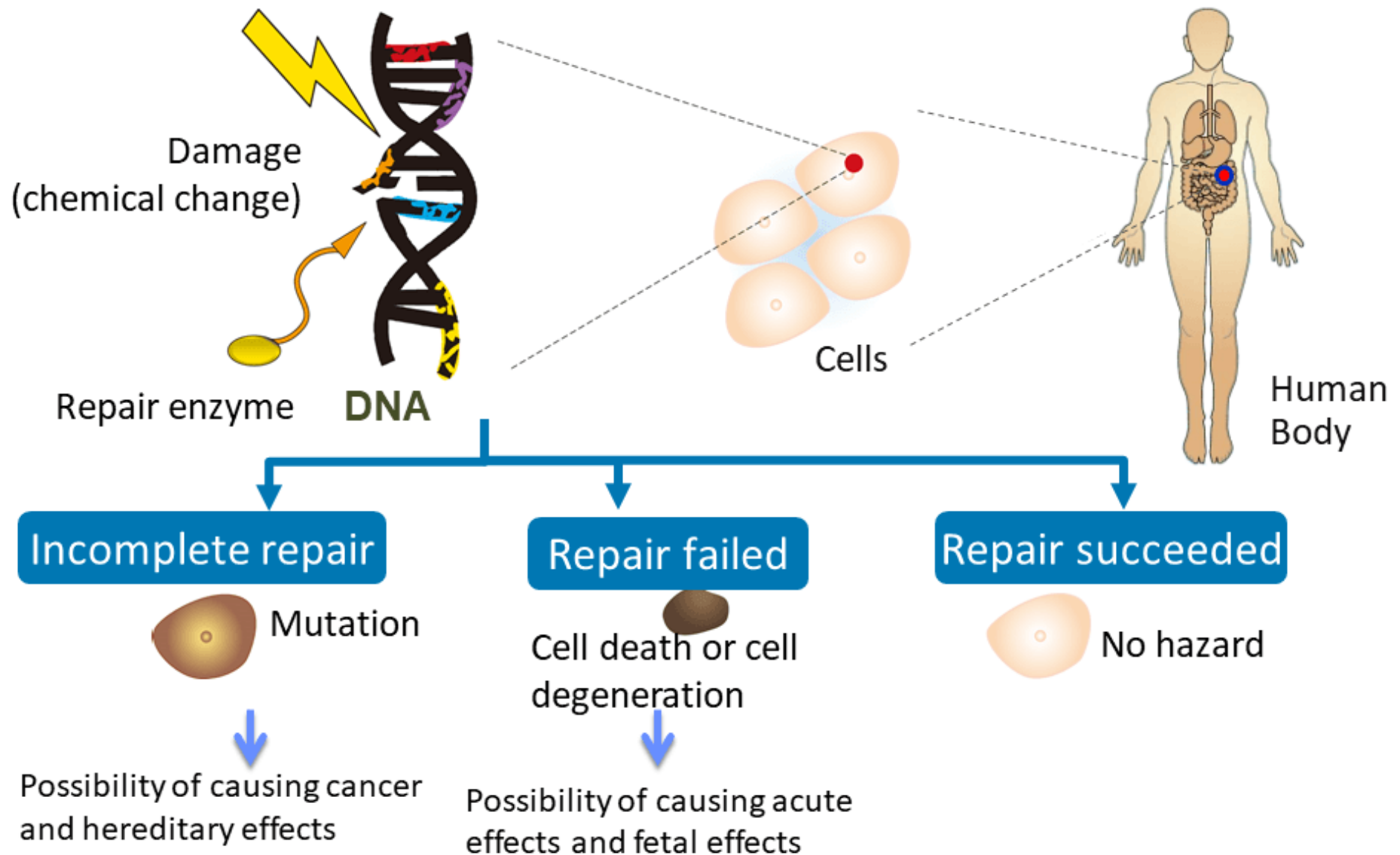
T_b : Biological half-life

Radioactive materials



Metabolism of Tritiated Water in the Human Body

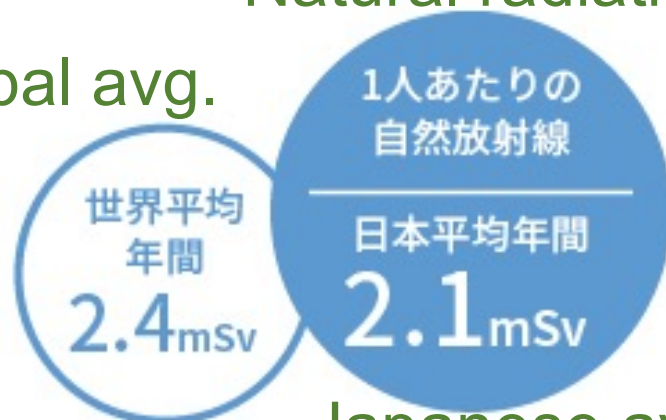




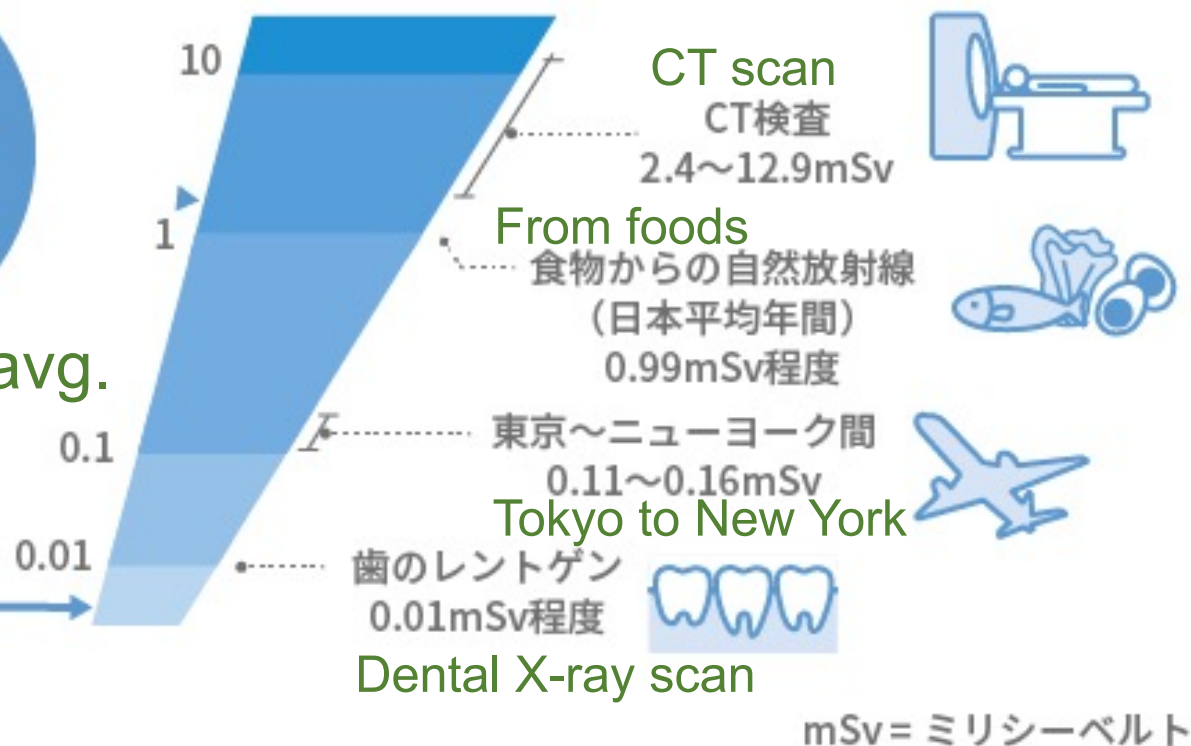
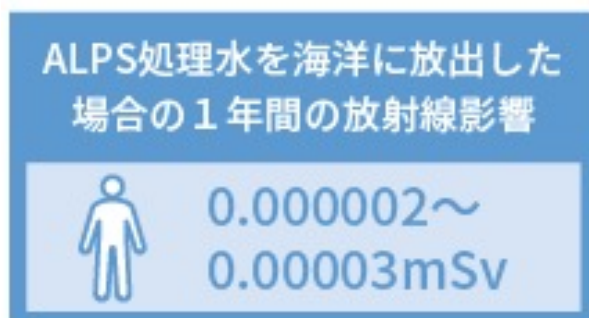
Comparison of Exposure Doses per Year

Natural radiation

Global avg.



Japanese avg.



Exposure dose when the ALPS-treated water is released into the sea

出典：国立研究開発法人量子科学技術研究開発機構放射線医学総合研究所の資料、環境省「放射線による健康影響等に関する統一的な基礎資料（令和2年度版）」第2章放射線による被ばくをもとに資源エネルギー庁にて作成

Breakdown of Natural Exposure Doses G-18 (Japanese)

Type of exposure	Breakdown of radiation sources	Effective dose (mSv/year)
External exposure	Cosmic rays	0.3
	Ground radiation	0.33
Internal exposure (inhalation)	Radon-222 (indoors and outdoors)	0.37
	Radon-220 (thoron) (indoors and outdoors)	0.09
	Smoking (Lead-210, Polonium-210, etc.)	0.01
	Others (uranium, etc.)	0.006
Internal exposure (ingestion)	Mainly Lead-210 and Polonium-210	0.80
	Tritium	0.0000082
	Carbon-14	0.01
	Potassium-40	0.18
Total		2.1

Source: Prepared based on "Environmental Radiation in Daily Life (2011)," Nuclear Safety Research Association

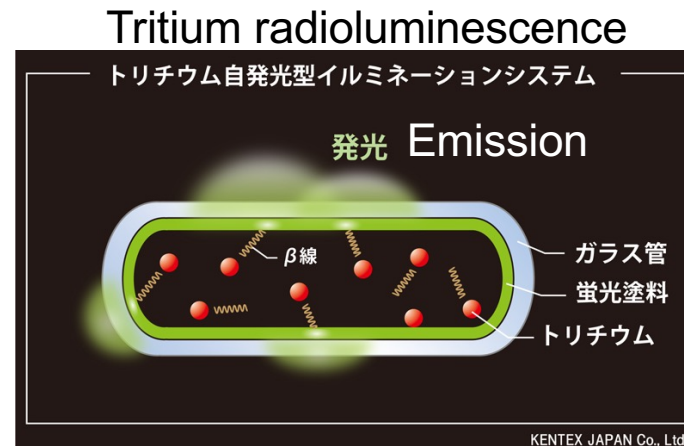
Source: BOOKLET to Provide Basic Information Regarding Health Effects of Radiation

Tritium Exposure Accidents

G-19

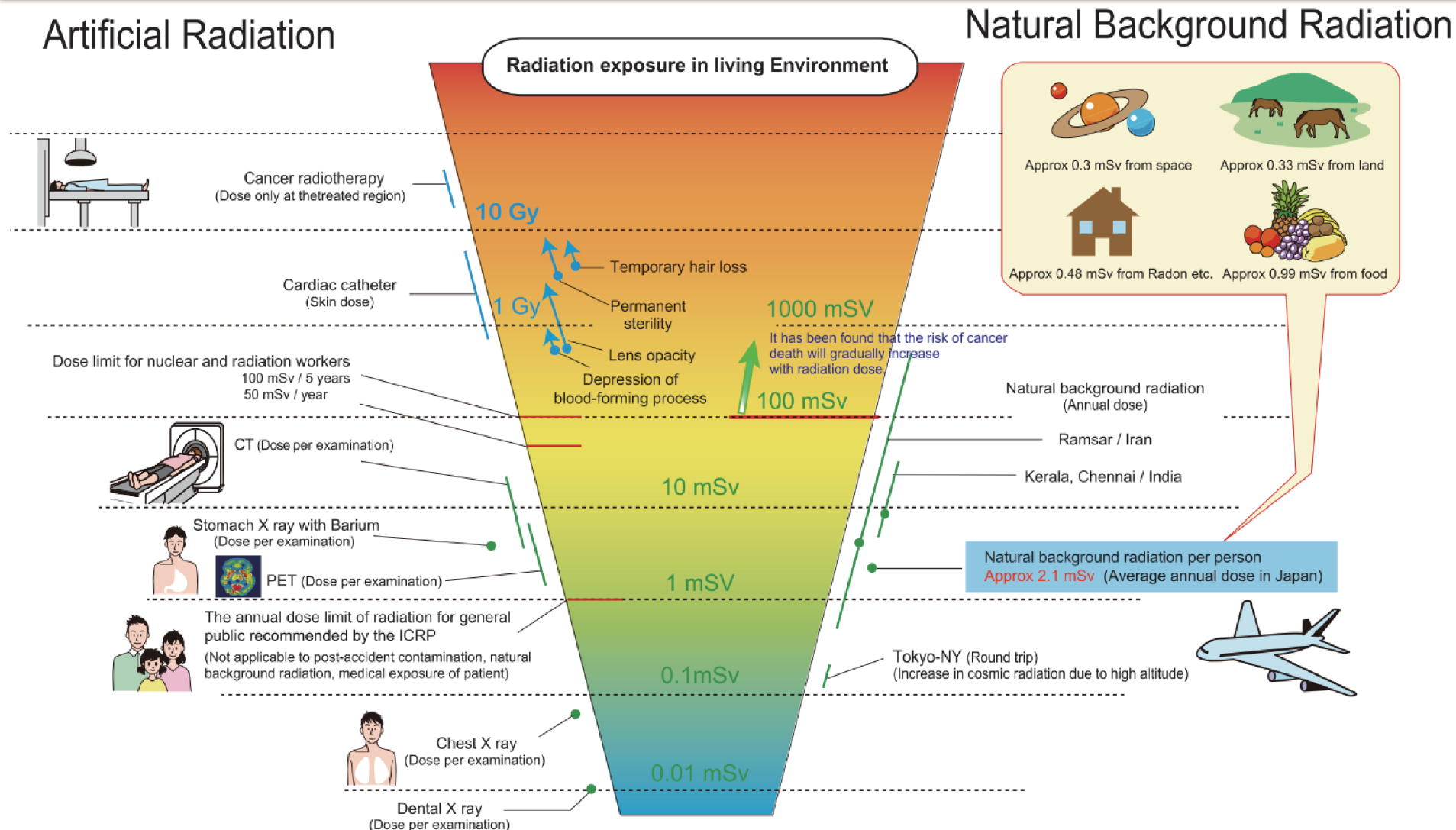
2 watch factories in Europe in the 1960s

- A factory worker ingested tritium over 7.4 years.
Exposure dose was estimated at **3-6 Sv**.
→ Developed isochromic anemia, and subsequently died of pancytopenia.
- A factory worker ingested tritium over 3 years.
Exposure dose was estimated at **10-20 Sv**.
→ Died of pancytopenia after following a similar disease as the same as above



Glass tube
Fluorescent paint
Tritium

Images are cited from KENTEX website (<https://www.kentex-jp.com>)



Sources:

- The 2008 UNSCEAR (United Nations Scientific Committee on the Effects of Atomic Radiation) Report
 - The 2007 ICRP (International Commission on Radiological Protection) Report
 - The exposure guideline of the Japan Association of Radiological Technologists
 - "Life Environmental Radiation (Calculation of the National Dose)," new edition
- Prepared by the National Institute of Radiological Sciences based on the sources above (May 2018)

mSv: millisieverts